## LETTERS

# Isotopic evidence for Mesoarchaean anoxia and changing atmospheric sulphur chemistry

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The evolution of the Earth's atmosphere is marked by a transition from an early atmosphere with very low oxygen content to one with an oxygen content within a few per cent of the present atmospheric level. Placing time constraints on this transition is of interest because it identifies the time when oxidative weathering became efficient, when ocean chemistry was transformed by delivery of oxygen and sulphate, and when a large part of Earth's ecology changed from anaerobic to aerobic<sup>1</sup>. The observation of non-mass-dependent sulphur isotope ratios in sedimentary rocks more than ~2.45 billion years (2.45 Gyr) old and the disappearance of this signal in younger sediments is taken as one of the strongest lines of evidence for the transition from an anoxic to an oxic atmosphere around 2.45 Gyr ago<sup>1-5</sup>. Detailed examination of the sulphur isotope record before 2.45 Gyr ago also reveals early and late periods of large amplitude non-mass-dependent signals bracketing an intervening period when the signal was attenuated<sup>5-9</sup>. Until recently, this record has been too sparse to allow interpretation, but collection of new data has prompted some workers<sup>8</sup> to argue that the Mesoarchaean interval (3.2-2.8 Gyr ago) lacks a non-mass-dependent signal, and records the effects of earlier and possibly permanent oxygenation of the Earth's atmosphere. Here we focus on the Mesoarchaean interval, and demonstrate preservation of a non-mass-dependent signal that differs from that of preceding and following periods in the Archaean. Our findings point to the persistence of an anoxic early atmosphere, and identify variability within the isotope record that suggests changes in pre-2.45-Gyr-ago atmospheric pathways for non-mass-dependent chemistry and in the ultraviolet transparency of an evolving early atmosphere.

The report seven years ago of non-mass-dependent isotope signals in a global distribution of Archaean and early Palaeoproterozoic sediments and metasediments<sup>5</sup> prompted a focused effort to understand the connections between the early sulphur cycle and the history of atmospheric oxygen. These non-mass-dependent effects do not follow the mass-dependent relationships  $\delta^{33}$   $\hat{S} \approx 0.5 \delta^{34}$   $\hat{S}$  and  $\delta^{36}$   $\hat{S} \approx$  $2\delta^{34}$ S, where the terms 0.5 and 2 reflect the relative mass differences between the numerator and denominator of <sup>33</sup>S/<sup>32</sup>S and <sup>36</sup>S/<sup>32</sup>S compared to those of <sup>34</sup>S/<sup>32</sup>S (see Methods for definitions of notation and a more detailed description of non-mass-dependent and massdependent isotope effects). Work on this topic has yielded a growing data set of  $\Delta^{33}$ S and  $\Delta^{36}$ S (a measure of the deviation from a reference array that approximates low temperature chemical equilibrium and passes through bulk Earth composition; see Methods) that is fully complementary to the existing  $\delta^{34}$ S data and interpretations made from this time series record. These data ( $\delta^{34}$ S,  $\Delta^{33}$ S and  $\Delta^{36}$ S) provide new insights into the processes and conditions necessary to produce non-mass-dependent sulphur isotope effects in the Archaean and earliest Palaeoproterozoic, as well as into the post-2.45-Gyr-ago world, where mass-dependent isotope effects are preserved<sup>7,10-12</sup>, and non-mass-dependent effects are largely absent—the exception being effects reported for some atmospheric samples and accumulations in ice cores<sup>13-15</sup>.

Figure 1 illustrates the dramatic change in the magnitude of  $\Delta^{33}$ S that occurred between ~2.45 and 2.32 Gyr ago<sup>1,6,12</sup>. The disappearance of large  $\Delta^{33}$ S in the rocks younger than ~2.45 Gyr reflects the suppression of the non-mass-dependent effects. Concomitant enhancement of mass-dependent signals (brought about by significantly larger sources of sulphate sulphur from oxidative weathering) reflects a shift from a sulphur cycle conducive to preservation of atmospheric isotope effects to a sulphur cycle in which biological isotope effects dominate. A corresponding increase in the range of  $\delta^{34}$ S (Fig. 2) has been attributed to the more significant role for sulphate reducing microorganisms in oceanic environments with higher sulphate concentrations. This figure also illustrates temporal variations in the Archaean  $\Delta^{33}$ S record, including an apparent



**Figure 1** | **Compilation of**  $\Delta^{33}$ **S versus age for rock samples.** Data are taken from refs 5–8, 12, 19–30. Ion probe data have been grouped by sample averages when measurements of different grains in the same sample did not vary outside reported uncertainties. The figure illustrates large  $\Delta^{33}$ S before ~2.45 Gyr ago (indicated by vertical line), and small but measurable  $\Delta^{33}$ S after 2.45 Gyr ago. Detailed examination of the pre-2.45-Gyr-ago record indicates an apparent minimum for  $\Delta^{33}$ S during the Mesoarchaean (2.8–3.2 Gyr ago). Uncertainties are smaller than the symbol sizes for  $\Delta^{33}$ S. Filled black circles are data from literature; filled grey circles are data presented in this study.

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The reports of non-mass-dependent sulphur isotope signals in modern atmospheric sulphate aerosols<sup>13</sup>, and in sulphate produced from volcanogenic SO<sub>2</sub> recovered from horizons in Antarctic ice cores and firn<sup>14,15</sup>, have been attributed to stratospheric reactions involving SO<sub>2</sub>, and provide one link to the atmosphere<sup>13–15</sup>. Additional links to the atmosphere have been made on the basis of photolysis experiments with SO<sub>2</sub>, a species inferred to be present in the atmosphere throughout Earth history. These experiments yield large non-mass-dependent isotope effects that exhibit a dependence on wavelength and are similar in many respects to the sulphur isotope compositions observed in the geologic record<sup>2</sup>.

The connection to these photolysis experiments and to modern stratospheric chemistry is taken as evidence that the early atmosphere was (partially or wholly) transparent to deep ultraviolet radiation— to allow similar chemical reactions to occur. An atmosphere transparent to deep ultraviolet would imply a low column depth for ozone, and also for oxygen because ozone is maintained at present high levels by chemistry involving atmospheric oxygen. Models of atmospheric chemistry imply a second connection to oxygen concentrations in the atmosphere. This is because the pathways for the transfer of nonmass-dependent sulphur to Earth's surface as aerosols (S<sub>8</sub> and sulphate) depend on whether oxygen and other oxygen species were high enough to consume neutral sulphur species (for example, S, S<sub>2</sub> and S<sub>3</sub>)<sup>3</sup>. In low oxygen conditions, S<sub>8</sub> aerosols are stabilized in addition to sulphate aerosols and this allows for a more efficient



Figure 2 | Plot of  $\Lambda^{33}$ S versus  $\delta^{34}$ S that illustrates the fields defined by preand post-2.45-Gyr-ago samples. The dotted grey line marks the limits of the observed post-2.45-Gyr-ago field. Fields are also included for published data, including samples from the Mozaan group<sup>7</sup> (filled black diamonds), samples from the Hardey Formation<sup>8</sup> (filled black circles), samples from the Mosquito Creek Formation<sup>8</sup> (filled black squares), and data from this study (filled grey squares). The field for the data from this study is outlined with a dashed black outline, as are the fields for literature data for the Hardey Formation, the Mosquito Creek Formation and the Mozaan Group, which are outlined with dotted black outlines and also shaded in grey. Grey unfilled triangles are literature data for post 2.45-Gyr-ago samples and grey filled circles circles are literature data for the data presented in this study are smaller than the symbol sizes for  $\Lambda^{33}$ S and  $\delta^{34}$ S.

transfer of atmospheric non-mass-dependent signals to surface sulphur and sulphate pools.

To investigate further the possibility of a non-mass-dependent effect in the Mesoarchaean (and slightly post-Mesoarchaean) record, we illustrate the covariation of  $\delta^{34}$ S with  $\Delta^{33}$ S for Mesoarchaean samples (Fig. 2), which provides a means to explore the significance of the attenuated  $\Delta^{33}$ S in this interval. Although the absolute magnitude of the  $\Delta^{33}$ S effect for these samples is smaller than observed in other portions of the Archaean (and in particular, the earliest Palaeoproterozoic), the  $\delta^{34}$ S versus  $\Delta^{33}$ S field is entirely consistent with the overall field for the Archaean/earliest Palaeoproterozoic data; it is also distinct from samples younger than 2.45 Gyr when the atmosphere was undeniably oxidized. We include data presented by Ohmoto *et al.*<sup>8</sup> and Ono *et al.*<sup>7</sup> in Fig. 2 to illustrate the striking consistency of each existing set of Mesoarchaean data with the non-mass-dependent (or pre-2.45-Gyr ago) array rather than the mass-dependent (or post-2.45-Gyr ago) array.

Although the non-mass-dependent signal may have been attenuated during the Mesoarchaean, our compilation illustrates a larger range of  $\Delta^{33}$ S variation than observed by Ohmoto *et al.*<sup>8</sup> and further suggests active non-mass-dependent chemistry. To support this contention, we provide an additional (and established) metric for distinguishing non-mass-dependent from mass-dependent signals with the concurrent measurement of <sup>36</sup>S abundances and the construction of  $\Delta^{36}$ S/ $\Delta^{33}$ S arrays (Fig. 3)<sup>5,7</sup>. The low  $\Delta^{36}$ S values in Fig. 3a reflect Rayleigh amplification of mass-dependent fractionation effects and define the limits of the mass-dependent field; they are well explained by theoretical considerations<sup>16</sup>. Covariation of  $\Delta^{36}$ S with  $\Delta^{33}$ S forms one of the principal arguments for a gas-phase photolytic origin of the effects observed in the Archaean/earliest Palaeoproterozoic record, and mass-dependent samples define a restricted field on a plot of  $\Delta^{36}$ S versus  $\Delta^{33}$ S that is distinct from the field defined by Archaean/earliest Palaeoproterozoic samples.

The Mesoarchaean data from this study, and one data point previously suggested to be anomalous by Ono et al.7, define fields that are distinct from the greater Archaean data set and the mass-dependent post-2.45-Gyr intervals. The differences between the Mesoarchaean data set and the data set for older Archaean samples are inconsistent with a non-mass-dependent signal that was inherited from preexisting terranes, and support the view of a pervasive anoxic atmosphere throughout the Archaean eon and into the earliest Palaeoproterozoic. Although some Mesoarchaean samples (for example, from the Nsuzi, Moodies and Mozaan<sup>7</sup> Groups) overlap the mass-dependent array for  $\Delta^{36}$ S/ $\Delta^{33}$ S, evidence for the large  $\delta^{34}$ S range that is required to produce these minor isotope variations by mass-dependent processes (see, for example, ref. 16) is not observed in samples from this time interval. We also point out that the Mesoarchaean data form distinct fields when grouped at the formation level (Fig. 3b), illustrating fundamental changes in Mesoarchaean non-mass-dependent signals, and we highlight similarities in the Mesoarchaean signals from sequences in South Africa and those in Western Australia.

Insight into the origin of the Mesoarchaean isotope variation (a diminished  $\Delta^{33}$ S amplitude and variable  $\Delta^{36}$ S/ $\Delta^{33}$ S) is provided by considering results from photochemical experiments and by comparisons with other Archaean intervals. The  $\delta^{34}$ S,  $\Delta^{33}$ S and  $\Delta^{36}$ S variations for some parts of the Archaean record—for example, the ~3.4-Gyr-old successions in Western Australia and South Africa appear to be well-matched by experiments undertaken with ~193 nm radiation<sup>2</sup>, and this observation led to the suggestion that deep-ultraviolet photolysis of sulphur dioxide provided an explanation for the origin of the non-mass-dependent signal in the Archaean eon. A subsequent study<sup>6</sup> documented a different relationship for  $\delta^{34}$ S and  $\Delta^{33}$ S variations (for the ~2.5-Gyr-old Mt McRae Shale), which fell between arrays produced by photolysis at 193 nm and experiments undertaken at longer wavelengths<sup>17</sup>: this led to a revised explanation for the non-mass-dependent signal—that it reflects a



- Australian Diamictite Upper Jeppestown
- Δ Mozaan Group
  - Nsuzi and Moodies Groups

combination of effects produced at shorter and longer ultraviolet wavelengths<sup>6,18</sup>. The data for the Mesoarchaean produce another relationship between  $\delta^{34}$ S,  $\Delta^{33}$ S and  $\Delta^{36}$ S that adds complexity but may also provide a way to reconcile the observations made of Palaeoarchean<sup>2</sup> and Neoarchaean<sup>6</sup> samples if they reflect changes in the source reactions that occurred through Archaean time.

The photolysis experiments<sup>2,18</sup> exhibit a wavelength dependence that can account for the observed  $\delta^{34}$ S,  $\Delta^{33}$ S and  $\Delta^{36}$ S relationships, and we speculate that changing transparency of the atmosphere may account for the observed minor sulphur isotope variation across the Archaean eon. In this context, the presence of variability for the Mesoarchaean data on plots of the  $\Delta^{36}$ S versus  $\Delta^{33}$ S (Fig. 3b) may also signal changes in atmospheric sulphur chemistry or transport. It is not clear whether the distinct arrays for samples grouped at

### Figure 3 | Plot of $\Delta^{36}$ S versus $\Delta^{33}$ S for mass-dependent and mass-

**independent data.** a, Plot of  $\Delta^{36}$ S versus  $\Delta^{33}$ S for pre-2.45-Gyr-ago samples (circles)<sup>5</sup> and for mass-dependent data (triangles) <sup>16,19,20</sup> (note that we have not included data from ref. 5 in the mass-dependent data set because of concerns about the exponent for  $\Delta^{36}$ S noted in ref. 21). **b**, Plot of  $\Delta^{36}$ S versus  $\Delta^{33}$ S for samples reported in this study and by Ono *et al.*<sup>7</sup>. The field defined by the Archaean/earliest Palaeoproterozoic samples is outlined by a dotted line, and the field defined by the mass-dependent samples is outlined by a dotted line and shaded light grey. Data grouped into the field outlined by the solid grey line include data from the Kameeldoorns Formation (filled grey diamonds), the Tumbiana and Kylena Formations (filled grey squares) and the Booysens Formation (filled grey circles). Data grouped into the field outlined by a solid black line includes the Mozaan Group (unfilled triangles), the Upper Jeppestown Formation (unfilled squares), the Middle and Upper Coronation Shale (unfilled diamonds) and the Promise Quartzite and Tillites (unfilled circles). Data for the Lower Jeppestown and Middle Government subgroup (filled black diamonds) and from the basal Fortescue group associated with diamictites (filled black squares) are grouped separately in a field surrounded by a dashed black line, and data for samples from the Nsuzi Group and the Moodies Group (filled black circles) are in a field outlined by a dashed grey line. Uncertainties for  $\Delta^{33}$ S are smaller than the symbol size, and uncertainties for  $\Delta^{36}$ S are twice the symbol size.

the formation level within the Mesoarchaean indicate that the nonmass-dependent reactions tell us about specific states of the atmosphere, but we suggest that the fanning of these arrays rather than a coherence from formation to formation points to a control that varies on timescales represented by the formations themselves. Our data therefore point to continued low oxygen conditions throughout the Archaean, and imply changes in chemistry. These changes might be related to variations in the relative abundance of trace gas species (maybe  $SO_2$ ), or to fluctuations in ultraviolet radiation transmitted through the atmosphere. Such fluctuations could reflect changes in atmospheric composition or the formation of atmospheric organic aerosol hazes.

#### METHODS SUMMARY

Sulphur isotope analyses were undertaken using standard fluorination methods and dual inlet gas-source mass spectrometry of sulphur extracted from samples by standard techniques (see Methods for details). Sulphur isotope ratios were measured for  $\mathrm{SF}_6$  as the analyte using a ThermoFinnigan MAT 253. The data are described using standard notation:  $\delta^{34}S = 1,000 \times (({}^{34}S/{}^{32}S)_{sample}/({}^{34}S/{}^{32}S)_{ref} - 1),$  $\Delta^{33}S = 1,000 \times (({}^{33}S/{}^{32}S)_{sample} / ({}^{33}S/{}^{32}S)_{ref} - (({}^{34}S/{}^{32}S)_{sample} / ({}^{34}S/{}^{32}S)_{ref})^{0.515}), and$  $\Delta^{36}S = 1,000 \times (({}^{36}S/{}^{32}S)_{sample} / ({}^{36}S/{}^{32}S)_{ref} - (({}^{34}S/{}^{32}S)_{sample} / ({}^{34}S/{}^{32}S)_{ref})^{1.9}).$ 

Full Methods and any associated references are available in the online version of the paper at www.nature.com/nature.

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#### **METHODS**

We use a framework for distinguishing non-mass-dependent and massdependent isotope effects made using data collected from the geologic record that is anchored in an understanding of the way that mass-dependent processes operate. Our framework is distinct from those that place the distinction between non-mass-dependent and mass-dependent fractionations using only a threshold of  $\Delta^{33}$ S (for example, at ~0.4%; ref. 31) because it also considers information present in correlations between  $\delta^{34}$ S,  $\Delta^{33}$ S and  $\Delta^{36}$ S. While this provides a more robust test of the presence of non-mass-dependent effects, we recognize that there will be an overlap in the fields for mass-dependent and non-massdependent signals. In some cases, samples that belong to non-mass-dependent populations may overlap with the range of the mass-dependent population.

In this study, sulphur from 66 samples were extracted using chromium reduction and converted to silver sulphide, converted to SF<sub>6</sub> by fluorination in a Ni reaction vessel at 250 °C with ~10× excess F<sub>2</sub> gas, and then purified by cryogenic and chromatographic techniques as described elsewhere<sup>32</sup>. Isotope ratios were determined by dual inlet IRMS using a ThermoFinnigan MAT 253 mass spectrometer with simultaneous monitoring of 127, 128, 129 and 131 (AMU/ $e^-$ ). Uncertainties on mass-dependent reference materials are better than ±0.2‰, ±0.01‰ and ±0.2‰ in  $\delta^{34}$ S,  $\Delta^{33}$ S and  $\Delta^{36}$ S, respectively. Uncertainties on the measurements reported here are estimated to be better than 0.2‰, ±0.02‰ and ±0.2‰. The results of our measurements are presented in Supplementary Table 1 and in Figs 1, 2 and 3.

Mass-dependent and non-mass-dependent fractionation. Geological interpretation of the sulphur isotope record has benefited significantly from the context provided by an understanding of the mechanisms that produce massdependent and non-mass-dependent isotopic fractionations. Mass-dependent fractionation effects arise when variations in chemical and physical properties of isotopologous species depend on the relative differences in mass of the atoms in a molecule, or the mass of the molecular species itself. Mass-dependent fractionations include, but are not limited to, equilibrium isotope exchange, diffusion, gravitational separation and biological effects<sup>32-36</sup>. Non-mass-dependent fractionations arise when variations in chemical and physical properties of molecular species depend on factors other than or in addition to the mass of the constituent isotopes<sup>37-39</sup>. Non-mass-dependent fractionations include those associated with symmetry changes of molecular species that arise from a particular isotope substitution, effects associated with internal conversions and intersystem crossings between bound and unbound states, effects associated with nuclear and electronic spin coupling, effects associated with self or mutual shielding, and effects that result from the physical production or destruction of an isotope. It has been established that non-mass-dependent and mass-dependent effects can occur together during chemical reactions or physical processes. For the purposes of this discussion, we will discuss non-mass-dependent effects as those that produce large deviations from mass-dependent fractionation arrays and involve factors other than mass, and we therefore do not include effects associated with mixing of different pools that are fractionated by mass-dependent processes. Our mass-dependent reference array is defined both by model predictions that take into consideration chemical and kinetic mass-dependent reactions and also by observations from oceanic sedimentary rocks. For this comparison, we have not included analyses of reagents or of samples of unknown provenance or age reported in refs 39 or 40.

We interpret the sulphur isotope data presented in Ohmoto *et al.*<sup>31</sup> to indicate that primary non-mass-dependent chemical effects were part of the sulphur cycles during deposition of the 2.76-Gyr-old Hardey Formation and the 2.92-Gyr-old Mosquito Creek Formation. These workers suggested that the signal observed in the rock record between 2.8 and 3.0 Gyr ago may reflect alternative processes, such as (1) high altitude chemistry involving sulphur dioxide photo-chemistry, (2) Rayleigh amplification of mass-dependent effects such as those that occur at fractionation crossovers, or (3) non-mass-dependent chemical reactions associated with the sulphate reduction reactions by amino acids. The first possibility has been discussed in ref. 41 and considered unlikely, considering

the distribution of volcanogenic sulphur in the atmosphere and constraints imposed by models of volcanic plume ascent, and the second suggestion can be ruled out on the basis of relationships between  $\Delta^{33}$ S and  $\Delta^{36}$ S that do not form a single array as is expected for Rayleigh amplification of a mass-dependent process, even one occurring at fractionation crossovers. The third suggestion remains to be demonstrated in a form that can be evaluated in the context of the emerging record of the four sulphur isotopes that includes evidence for systematic changes of  $\Delta^{36}$ S/ $\Delta^{33}$ S, and possibly of systematic changes in the magnitude of the  $\Delta^{33}$ S and the relationship between  $\delta^{34}$ S and  $\Delta^{33}$ S. The viability of alternative hypotheses must be demonstrated to be consistent with the  $\Delta^{36}$ S/ $\Delta^{33}$ S and  $\delta^{34}$ S/ $\Delta^{33}$ S. It also should account for any other observations in the record, including the possibility that there may be basinal and global correlations in the record and systematic variations in the relationships between these various measures of sulphur isotope composition.

Our conclusion primarily differs from those in Ohmoto et al.<sup>31</sup> because we have included constraints implied by the  $\delta^{34}$ S data, which was not used to make the assertions in Ohmoto *et al.*<sup>31</sup>. This evaluation can be tested further using  $\Delta^{36}$ S and the relationships between  $\Delta^{36}S$  and  $\Delta^{33}S$ . For instance, it is possible that some of the  $\Delta^{36}$ S/ $\Delta^{33}$ S variation in the Archaean data set may reflect the convolution of a mass-dependent signal with a non-mass-dependent signal of Archaean origin, and this possibly accounts for the variation in the  $\Delta^{36}S$  at  $\Delta^{33}$ S = 0 for the larger Archaean data set, but this explanation does not account for the full magnitude of the  $\Delta^{36}$ S variability in the Mesoarchaean and changes in  $\Delta^{36}$ S/ $\Delta^{33}$ S. We also recognize the possibility that processes other than photochemistry may contribute to the signals that are observed, but note that experimental evidence for non-mass-dependent effects involving gases of volcanogenic origin in combination with constraints from chemical models of anoxic atmospheres anticipates these effects in Earth's earliest anoxic environments. A subtle, but important point here is that the sulphur signal cannot be simply be an inherited detrital component if it has a different multiple sulphur isotope character, and even if there are detrital components, an authigenic Mesoarchaean component is inferred. An unresolved aspect of this signal in the Archaean sections is the observation of larger non-mass-dependent signals in shales, banded iron formations, and barites when compared to coarser clastic sediments.

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